Chemical Synthesis of Cytidine-5'-monophosphono-N-acetylneuraminic Acid (CMP-Neu5Ac)

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Abstract: CMP-Neu5Ac was synthesized for the first time from the cytidine 5'-phosphoramidite and the Neu5Ac derivative bearing allyl and allyloxycarbonyl groups as the protecting groups.

CMP-Neu5Ac 6 is known to be a substrate of sialyltransferase, which transfers N-acetylneuraminic acid $(Neu5Ac)^1$ to oligosaccharides, glycolipids and glycoproteins.² CMP-Neu5Ac has been enzymatically synthesized using CMP-Neu5Ac synthetase.³ But, this method can hardly be applied for CMP-Neu5Ac derivatives,⁴ because CMP-Neu5Ac synthetase strictly recognizes the substrates. Therefore, the chemical synthesis is the most promising method for the preparations of CMP-Neu5Ac derivatives. In the chemical synthesis, there are two problems. One problem is the instability of CMP-Neu5Ac under both basic and acidic conditions. Another is the low reactivity of the hindered anomeric hydroxyl group of Neu5Ac for the phosphorylation. Very recently, Wong *et al.*⁵ have synthesized a fully protected CMP-Neu5Ac derivative via the phosphoramidite method. However, they did not perform deprotection. In this paper, we wish to report the synthesis of CMP-Neu5Ac by the phosphoramidite method employing allyl and allyloxycarbonyl (AOC)⁶ groups as the protecting groups which can be removed smoothly by treatment with palladium(0) catalyst under neutral conditions.⁷

Both the methoxy and acetyl groups could be removed from methyl [2-(trimethylsilyl)ethyl 5-acetamido-4,7,8,9-tetra-O-acetyl-3,5-dideoxy-D-glycero-a-D-galacto-2-nonulopyranosid]onate⁸ 1 under alkaline conditions, and subsequently, the carboxyl group was protected as the allyl ester to give 2 in 90% yield according to the literature procedure.⁹ AOC groups at 4, 7, 8 and 9 positions could be successfully introduced to 2 with 50 equiv each of AOC-tetrazolide and 2,6-lutidine in a very small amount of DMF at room temperature for 24 h to give 3 in 94% yield.¹⁰ Removal of the trimethylsilylethyl (SE) group from 3 was carried out by treatment with trifluoroacetic acid¹¹ at room temperature for 30 min to give 4 in 97% yield. 4 was treated with 6 equiv of an appropriately protected cytidine 5'-phosphoramidite 7¹² in the presence of 15 equiv of tetrazole at 0 °C for 10 min. Successive addition of triethylamine could stabilize the phosphite intermediate.¹³ Oxidation of the phosphite was performed by *tert*-butyl hydroperoxide.¹⁴







Scheme 1. (a) (i) NaOCH₃ / CH₃OH; (ii) NaOH / H₂O; (iii) Dowex 50Wx8 (H⁺ form); (iv) Cs₂CO₃; (v) allyl bromide (8 equiv) in DMF. (b) AOC-tetrazolide (50 equiv) / 2,6-lutidine (50 equiv) in DMF. (c) trifluoroacetic acid. (d) (i) compound 7 (6 equiv), tetrazole (15 equiv) in CH₃CN (8ml) at 0°C; (ii) Et₃N (30 equiv), 80% tBuOOH (20 equiv), at 0 °C to room temperature. (e) (i) HCOO⁻ nBuNH₃⁺ (60 equiv); (ii) triphenylphosphine (1.5 equiv) / tetrakis(triphenylphosphine) palladium(8) (3 equiv) in THF (3ml).

The structure of 5 was supported by ³¹P NMR.¹⁵ Isolation of 5 was tried by silica gel column chromatography, but could not be achieved due to the instability of 5 bearing the protecting groups. Therefore, 5 was immediately deprotected by addition of a solution of triphenylphosphine and tetrakis(triphenylphosphine)palladium(0) in tetrahydrofuran. Purification of the reaction mixture by Sephadex G-15 column chromatography¹⁶ gave 6 in 25% yield (14mg) based on 4. Structure of 6 was supported by its mobility on TLC, and confirmed by ¹H and ³¹P NMR spectra.¹⁷ The ¹H NMR spectrum (Figure 1) was identical with that of the enzymatically synthesized authentic sample.



Figure 1 ¹H NMR spectrum of CMP-Neu5Ac (nBuNH₃⁺ salt)

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References and Notes

- 1) Schauer, R. Adv. Carbohydr. Chem. Biochem. 1982, 40, 131-234.
- 2) Fisherman, P. H.; Brady, R. O. Science 1976, 194, 906-915.
- 3) Haverkamp, J.; Beau, J. M.; Schauer, R. Z. Physiol. Chem. 1979, 360, 159-166.
- Liu, L.- C.; Shen, G.- J., Ichikawa, Y.; Rutan, J. F.; Zappa, G. Z.; Vann, W. F.; Wong, C.- H. J. Am. Chem. Soc. 1992, 114, 3901-3910.
- 5) Kondo, H.; Ichikawa, Y.; Wong, C.- H. J. Am. Chem. Soc. 1992, 114, 8748-8750.
- 6) Tsuji, J. Acc. Chem. Res. 1973, 6, 8-15.

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- Hayakaya, Y.; Kato, H.; Uchiyama, M.; Kajino, H.; Noyori, R. J. Org. Chem. 1986, 51, 2400-2402.
- 8) Hasegawa, A.; Ito, Y.; Ishida, H.; Kiso, M. J. Carbohydr. Chem. 1989, 8, 125-133.
- Furuhara, K.; Sato, S.; Anazawa, K.; Goto, M.; Takayanagi, H.; Ogura, H. Chem. Pharm. Bull. 1987, 35, 3609-3614.
- ¹H NMR (d, ppm, CDCl₃) 1.92 (1H, s, NAc), 2.28-2.36 (2H, m, H-3ax and H-3eq), 4.22 (1 H, d, J=10.6 Hz, H-5), 4.28 (1H, d, J=9.2 Hz, H-9'), 4.38 (1H, d, J=10.5 Hz, H-6), 4.53-4.73 (11H, m, H-9 and allyl), 5.10-5.43 (13H, m, H-8, H-4, H-7 and allyl), 5.83-6.02 (5H, m, allyl), 6.14 (1H, d, J=9.6 Hz, NH).
 ¹³C NMR (d, ppm, CDCl₃) 23.3 (NAc), 36.0 (C-3), 49.5 (C-5), 65.7 (C-9), 67.0, 68.7, 68.7, 69.1, 69.4 (allyl), 70.5 (C-6), 72.0 (C-7), 73.3 (C-4), 74.8 (C-8), 95.0 (C-2), 118.9, 118.9, 119.2, 119.3, 119.3 (allyl), 130.9, 131.2, 131.3, 131.5, 131.6 (allyl), 154.2, 154.4, 154.6, 154.9 (carbonate),

The chemical shifts were assigned by H-H COSY and H-C COSY.

- 11) Jansson, K.; Frejd, T.; Kihlberg, J.; Magnusson, G. Tetrahedron Lett. 1988, 29, 361-362.
- 12) Havakawa, Y.; Hirose, M.; Novori, R. Nucleosides & Nucleotides 1989, 8, 867-870.
- 13) ³¹P NMR before oxidation (d, ppm, CH₃CN-CDCl₃, 1:1, v/v) 136.5, 135.1.
- 14) Havakawa, Y.; Uchiyama, M.; Novori, R. Tetrahedron Lett. 1986, 27, 4191-4194.
- 15) ³¹P NMR after oxidation (d, ppm, CH₃CN-CDCl₃, 1:1, v/v) -5.7, -6.4.
- 16) Thiem, J.; Treder, W. Angew. Chem., Int. Ed. Engl. 1986, 25, 1096-1097.
- 17) ³¹P NMR (d, ppm, D₂O) -4.0 R_F=0.2. The sample was loaded on HPTLC-Fertigplatten Cellulose F_{254S}(Merck) and developed with ethanol-1 M ammonium acetate (7:3, v/v). Then, cytidine component was detected by UV light and N-acetylneuraminic acid component was detected by the periodate resorcinol method.¹⁸
- 18) Jourdian, G. W.; Dean, L.; Roseman, S. J. Biol. Chem. 1971, 246, 430-435.

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168.0 (NCO), 171.0 (C-1).